

Hydrogen Peroxide in Urine as a Potential Biomarker of Whole Body Oxidative Stress

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The level of hydrogen peroxide (H₂O₂) in urine has been suggested as a potential biomarker of whole body oxidative stress, but issues of stability, reproducibility and biological variation have not been investigated to date. In this study, we used a refined protocol, which demonstrated improved sensitivity and precision, to determine the stability of H_2O_2 in urine, and to measure its concentration in apparently healthy subjects. We also investigated intra-individual variation within and between days. Results showed that H_2O_2 in urine is stable for up to 48 h at 4°C, however, storage of urine at room temperature was associated with up to 50% increase in H_2O_2 concentration over a few hours. Total H_2O_2 in freshly voided urine from 55 healthy, fasting subjects ranged from 0.84 to 5.71 μM, or 90-1164 μmol H_2O_2 /mol creatinine. Intra-individual variation was wide. Even when concentration corrected and collected at the same time of day, 2- to 3-fold variation was seen over 4 consecutive days, and over the course of a single day the creatinine-corrected H₂O₂ also varied significantly. We suggest that this large biological variation limits the usefulness of urine H_2O_2 as a biomarker of oxidative stress, the exception being when the effects of disease, therapy or diet induce very large changes in its concentration.

Keywords: Hydrogen peroxide; Urine; Oxidative stress; ROS

INTRODUCTION

There are various biological markers that are used to assess oxidative stress, including F2 isoprostanes, 8-oxo-dG, and protein carbonyls. These biomarkers give a view of oxidative damage to, respectively, lipid, DNA and protein, but individually do not give a "global" view of whole body stress. [1–5] Furthermore, these tests are expensive and difficult to perform well. A simple and inexpensive test of overall oxidative

stress would be a useful addition to the test menu for researchers investigating oxidative stress and diet in relation to ageing and chronic degenerative disease. In this regard, it has been suggested that the level of hydrogen peroxide (H_2O_2) in urine may be a possible candidate for a biomarker of the extent of whole body oxidative stress. [6-8] This is an attractive concept, as urine is easy to obtain, and H_2O_2 is simple to measure.

Before assessing the effect of dietary change, lifestyle, ageing or disease on a putative biomarker, however, its biological, or "normal", variation must be determined within- and between individuals, and the stability of the biomarker in samples ex vivo must be determined.^[9] Our primary aims in this study, therefore, were to investigate inter- and intraindividual variation in urine H₂O₂ concentration in healthy subjects and to determine the effect of urine storage at room temperature and at 4°C. We employed a commonly used, technically simple method for H₂O₂ measurement, often referred to as the FOX assay, [8-12] which employs ferrous ion oxidation by hydrogen peroxide, with subsequent formation of a blue-violet coloured Fe²⁺-xylenol orange chromogen. Because the assay showed rather poor precision and sensitivity in our preliminary studies, we aimed also to refine the method to lower the limit of detection and increase the precision.

MATERIALS AND METHODS

All chemicals and reagents were of analytical grade. To prepare the working FOX reagent,



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1 ml of a 25 mM aqueous solution of ferrous ammonium sulphate FeSO₄(NH₄)₂SO₄·6H₂O (Sigma, St. Louis, MO, USA) in 2.5 M H₂SO₄ (Merck Chemical Co. Ltd, Darmstadt, Germany) was mixed with 100 ml of an aqueous solution of $125 \,\mu\text{M}$ xylenol orange (3,3-bis[N,N-Di(carboxymethyl)aminomethyl]-o-cresol-sulphonaphthalein; Sigma) containing 100 mM Sorbitol (>99% pure, BDH Laboratory Supplies, Poole, England). This solution was prepared just before use. A 30% H₂O₂ stock solution (Sigma) was used to prepare working H2O2 standards over the range 0-15 μM. Working standards (0, 2, 4, 6, 8, 10, 12 and 15 µM) were prepared freshly as needed from the stock 30% solution and using MilliQ ultrapure water (at least $18.2 \,\mathrm{M}\Omega$, prepared using a Millipore Ultrapure Water System, Millipore Corp., Bedford, MA). The working standards, controls and urine samples were treated identically using two protocols. One protocol followed the published FOX assay, as used in a commercially available kit method (Pierce Chemical Company, Rockford, USA) as follows: 20 µl of standards, control or urine was added to 200 µl of reagent (freshly prepared as described above) in a microplate well, mixed and left for 20 min at room temperature, and the absorbance at 590 nm was read against a reagent blank. The other protocol was based on this, however, in an attempt to improve the sensitivity of the assay we increased the sample:working reagent ratio, adding 100 µl sample to 170 μl reagent in a microplate well. Furthermore, because we noted that colour development had not stabilised after the 20 min reaction time recommended, we extended the incubation time after sample/reagent mixing to 60 min, by which time colour development was approaching a plateau value. Both protocols were performed in parallel at room temperature, using flat bottomed microtitre plate wells (Thermo LabSystems, Franklin, MA) and the same standards, controls and urine samples. Absorbance readings were taken using a TECAN SPECTRAFluor Plus microplate reader (TECAN Austria GmbH, Grodig, Austria). The limit of detection for each protocol was determined by calculating the mean + 3SD absorbance value of a blank (MilliQ water) measured 20 times by each protocol, and reading the corresponding value from the appropriate calibration line. Precision, in terms of in-run and between-run coefficients of variation (CV), was determined using freshly prepared solutions of H₂O₂. All urine samples were performed in duplicate by each protocol, and the mean value used.

To investigate inter- and intra- individual variation in urine H₂O₂, mid-stream urine samples were collected, into plain containers, from apparently healthy volunteers aged 20–55 years. Fasting urines from 55 volunteers (29 men, 26 women) were collected between 9 and 10 a.m. to investigate interindividual variation. In a separate experiment, 17 volunteers provided mid-stream urine samples at 9 a.m. (fasting), 12 midday, 2 p.m. and 4 p.m. on the same day. In a third experiment, 15 volunteers provided a mid-stream urine sample on each of 4 consecutive days: the time at which the urine was collected was the same for each subject on each of the 4 days, however, nine volunteers collected urine each morning at 10 a.m., while six collected urine at 2 p.m. each day. Freshly voided urines were immediately centrifuged at 4°C for 8 min at $2500 \, \text{rpm}$, and H_2O_2 measurements were performed immediately thereafter, using the refined protocol. Aliquots of centrifuged fasting urines were stored at 4°C for evaluating the effect of up to 48 h storage. In a separate study, the effect of up to 6 h storage at room temperature compared to 4°C was also investigated.

H₂O₂ results were concentration corrected and expressed as μmol H₂O₂ per mol creatinine, as well as in μ M total H₂O₂. Creatinine in urine (diluted in water as appropriate) was measured by an automated commercial creatinine assay kit (Biosystems Reagents and Instruments, Barcelona, Spain), which is based on the method of Jaffe. [13]

Ethical approval for the study was obtained from the University Ethics Sub-Committee, and all procedures involving human subjects complied with the Declaration of Helsinki as revised in 2000.

Statistical analysis: GraphPad Prism version 3.0 for Windows (GraphPad Software, San Diego, CA) was used to perform the Friedman test with Dunnett's post test for intra-individual within-day and between-day variation. The paired t-test was used to assess ex vivo changes. Statistical significance was sought at the 5% level.

RESULTS

After increasing the sample: reagent ratio and extending the incubation time from 20 to 60 min, the limit of detection was around 3-fold lower, and precision and sensitivity were markedly improved. In-run CVs for the original and refined protocols were, respectively, 3.37 and 1.99% (at $5.0 \,\mu\text{M} \, \text{H}_2\text{O}_2$; n = 11 in each case). Between-run CVs were, respectively, 4.98 and 2.17% at a similar H₂O₂ concentration (n = 9 in each case). The limits of detection of H₂O₂ with the original and refined protocols were, respectively, 2.5 and 0.8 μM. Figure 1 shows the dose-response curves obtained with the original protocol (squares) and with the refined protocol (circles). It can be seen that the sensitivity (gradient of the dose-response line) is almost 4-fold



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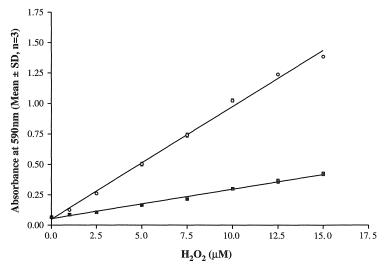


FIGURE 1 Dose-response (calibration) lines for the original (squares) and refined (circles) protocols for H₂O₂ measurement.

better using the refined protocol. Linearity was maintained at least to $12.5 \,\mu\text{M} \,\text{H}_2\text{O}_2$.

In regard to ex vivo changes, urine stored at 4°C for up to 48 h showed no significant changes (P >0.05) in H_2O_2 concentration: Mean (SD) at zero time, +5, +24 and +48 h were, respectively, 2.57 (1.19), 2.74 (1.48), 2.70 (1.38) and $2.63 (1.48) \mu M (n = 55 in)$ each case). Results of six urine samples that were stored and measured, in parallel, at room temperature and 4°C showed that with storage at room temperature, urine H₂O₂ increased by around 40% over 2h, and by 50% by 4h, but again no appreciable difference was seen in urines stored at 4°C (results not shown).

In regard to biological variation, total and creatinine-corrected H₂O₂ results on fasting urine samples (n = 55) are shown in Table I. Urine total H_2O_2 concentrations ranged from 0.84 to 5.71 μ M; concentration-corrected levels ranged from 90 to $1164 \,\mu\text{mol}$ H₂O₂/mol creatinine. There was no significant difference between urine H₂O₂ concentrations in men and women. There was a significant correlation between H₂O₂ concentration and creatinine concentration (r = 0.531; P < 0.0001), however, scatter was wide, indicating that differences in H₂O₂ concentration were not a simple function of urine concentration per se. Within-day intra-individual variation is shown in Fig. 2. From these data

TABLE I Urine H₂O₂ concentrations in healthy, fasting adults (Mean; Median (SD); n = 55)

	Total H ₂ O ₂ (μΜ)	Concentration corrected (µmol H ₂ O ₂ /mol creatinine)
All $(n = 55)$	2.57; 2.34 (1.19)	322; 229 (233)
Men $(n = 29)$	2.71; 2.34 (1.17)	346; 265 (266)
Women $(n = 26)$	2.42; 2.34 (1.22)	301; 214 (200)

it can be seen that the creatinine-corrected H₂O₂ concentration increased significantly (P < 0.05)during the day in both men and women, when compared to the fasting (9 a.m.) sample, with highest values seen in samples collected in the early afternoon (2 p.m.). Individual, between-day variation in 15 subjects is shown in Fig. 3. Results indicate that intra-individual variation in some subjects was very wide.

DISCUSSION

The use of biomarkers indicating oxidative stress is an important issue.^[9] In this study we used a refined protocol, which demonstrated improved sensitivity and precision, to determine the stability of H_2O_2 in urine, and the concentration in fasting urine of 55 apparently healthy subjects in order to investigate whether the level of H_2O_2 in urine is a suitable potential biomarker for use in studies of oxidative stress and the effect of age, disease, and diet on this. We also investigated the effect of sampling time and between-day intra-individual variation. Results showed that $\mathrm{H_2O_2}$ in urine is stable for up to $48\,\mathrm{h}$ at 4°C, however, storage of urine at room temperature was associated with up to 50% increases in H_2O_2 concentration over a few hours. This increase is presumably owing to autoxidation of urine constituents. This increase was seen also in a study by Lee et al. who showed increases of 17-139% in seven urine samples stored for 3.5 h at room temperature.[11]

In this current study, the concentrations of total H_2O_2 in freshly voided urine in 55 healthy subjects were lower than those previously reported. [7-8,11,14] The study by Varma and Devamanoharan,[14]



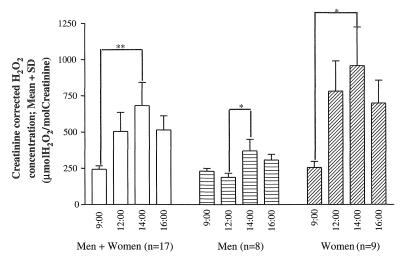


FIGURE 2 Creatinine-corrected H₂O₂ concentrations in urine of 17 healthy subjects (eight men [horizontal bars] and nine women [diagonal bars]); concentrations increased significantly (*P < 0.05; **P < 0.01) during the day.

which used a different method of measurement, reported mean (SD) urine total H₂O₂ in men and women of, respectively, 106(64) and 89(54) µM, and the studies by Lee, Halliwell and co-workers reported total levels of 0.4-109 µM in 20 healthy subjects. [7-8] In this current study, we collected urine from 55 fasting subjects, and no sample had a total H_2O_2 concentration above $6 \mu M$. Furthermore, values in men and women were not significantly different. In terms of creatininecorrected H₂O₂ levels in urine, recalculation of the only published data we are aware of [8] shows that spot urine concentrations in healthy, nonfasting subjects ranged from 276 to 1844 µmol/mol (n = 4). These values are similar to those found in this current, larger study (90-1164 µmol/mol), and again indicate a very wide inter-individual variation. Intra-individual variation in urine H₂O₂ concentrations is also large, as shown clearly in this study. Even when concentration-corrected and collected at the same time of day, 2- to 3-fold

variation was seen over 4 consecutive days, and over the course of a single day, the creatinine-corrected H₂O₂ varied significantly. This was most likely owing to intake of food containing components that generate H₂O₂, but may have been related also to physical activity and electrolyte balance. [15] There is a report that drinking coffee, a beverage that has been shown^[10] to generate significant amounts of H₂O₂, increased the urine H₂O₂ within 60 min in three of the four subjects studied, [8] and exercise and salt loading are also reported to affect H₂O₂ excretion.^[15]

In conclusion, these new data show that H₂O₂ can be measured simply and reproducibly in human urine by the refined protocol used, and that H_2O_2 is stable in urine at 4°C for up to 48 h. The concentration varies widely, however, both within- and between day, and this, in combination with the wide interindividual variation seen, severely limits the usefulness of urinary H₂O₂ as a potential biomarker for whole body oxidative stress unless the effects of

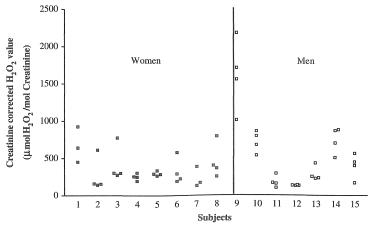


FIGURE 3 Creatinine-corrected H₂O₂ concentrations in urine of 15 healthy subjects (seven men [open squares] and eight women [closed squares]) over the course of 4 consecutive days; in each subject urine was collected at the same time each day.



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disease or therapy induce very large changes in its concentration.

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References

- [1] Halliwell, B. and Gutteridge, J.M.C. (1999) Free Radicals in Biology and Medicine (Oxford University Press, New York).
- [2] O'Reilly, J.D., Mallet, A.I., McAnlis, G.T., Young, I.S., Halliwell, B., Sanders, T.A. and Wiseman, H. (2001) "Consumption of flavonoids in onions and black tea: lack of effect on F2-isoprostanes and autoantibodies to oxidized LDL in healthy humans", Am. J. Clin. Nutr. 73, 1040–1044.
- [3] Wubert, J., Reder, E., Kaser, A., Weber, P.C. and Lorenz, R.L. (1997) "Simultaneous solid phase extraction, derivatization, and gas chromatographic mass spectrometric quantification of thromboxane and prostacyclin metabolites, prostaglandins, and isoprostanes in urine", Anal. Chem. 69, 2143–2146.
- Yoshida, R., Ogawa, Y. and Kasai, H. (2002) "Urinary 8-oxo-7, 8-dihydro-2'-deoxyguanosine values measured by an ELISA correlated well with measurements by high-performance liquid chromatography with electrochemical detection", Cancer Epidemiol. Biomark. Prev. 11, 1076-1081.
- Stadtman, E.R. and Levine, R.L. (2000) "Protein oxidation", Ann. NY Acad. Sci. 899, 191-208.

[6] Halliwell, B., Marie, V.C. and Lee, H.L. (2000) "Hydrogen peroxide in the human body", FEBS 486, 10–13.

- [7] Halliwell, B., Marie, V.C. and Ramalingam, J. (2000) "Hydrogen peroxide. Ubiquitous in cell culture and in vivo?", Life 50, 251-257.
- [8] Lee, H.L. and Halliwell, B. (2000) "Coffee drinking increases levels of urinary hydrogen peroxide detected in healthy volunteers", Free Radic. Res. 32, 463-467.
- [9] Griffiths, H.R., Moller, L., Bartosz, G., Bast, A., Bertoni-Freddari, C., Collins, A., Cooke, M., Coolen, S., Haenen, G., Hoberg, A.-M., Loft, S., Lunec, J., Olinski, R., Parry, J., Pompella, A., Poulsen, H., Verhagen, H. and Astley, S.B. (2002) "Biomarkers", Mol. Asp. Med. 23,
- [10] Lee, H.L., Ng, A.B.L. and Teng, F.Y.H. (1999) "Generation of hydrogen peroxide by 'antioxidant' beverages and the effect of milk addition. Is cocoa the best beverage?", Free Radic. Res.
- [11] Lee, H.L., Evans, P.J. and Halliwell, B. (1999) "Hydrogen peroxide in human urine: implications for antioxidant defense and redox regulation", Biochem. Biophys. Res. Commun. 262, 605-609.
- [12] Craig, G. and Janusz, M.G. (2000) "A critical evaluation of the effect of sorbitol on the ferric-xylenol orange hydroperoxide assay", Anal. Biochem. 284, 217-220.
- [13] Fabiny, D.L. and Ertingshausen, G. (1971) "Automated reaction-rate method for determination of serum creatinine with the CentrifiChem", Clin. Chem. 17, 696 - 700
- [14] Varma, S.D. and Devamanoharan, P.S. (1990) "Excretion of hydrogen peroxide in human urine", Free Radic. Res. Commun.
- [15] Kuge, N., Kohzuki, M. and Sato, T. (1999) "Relation between natriuresis and urinary excretion of hydrogen peroxide", Free Radic. Res. 30, 119-123.

